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Lesson 3.1 Introduction to Atoms Vocabulary: Atom - Electron - Nucleus - Proton - Energy level - Neutron - Atomic number - Isotope - Mass number -. Atomic Theory Studying atoms is difficult because atoms are so small. The smallest visible speck of dust may contain about 10 million billion atoms.

## **Lesson 3.1 Introduction to Atoms Vocabulary**

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## **Intro to Atoms reading and review - MISS SCHAEFER'S ...**

The basic particles of matter are called atoms. Atoms are very small and can form chemical bonds with other atoms. 3. Sample: I wanted to include more about atoms, but I ran out of room.

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## **Introduction to Matter - Bridgeway**

Introduction In this lesson, you will learn more about the atom's basic structure and the positive and negative charges of its subparticles. This lesson lays the groundwork for further study of static and current electricity by focusing on the idea of positive and negative charges at the atomic level.

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## **Introducing Atoms - Science NetLinks**

Pearson Chapter 1 - Introduction to Matter. Lesson 1: Describing Matter Lesson 2: Classifying Matter Lesson 3: Measuring Matter. ... two or more atoms put together to form a substance. ... Purple Pearson Chemistry - Chapter 5 - Chemical Reactions. 7 terms. Chapter 3 -- Elements and The Periodic Table - FLS ...

## **Pearson Chapter 1 - Introduction to Matter Flashcards ...**

Introduction to Atoms (pp. 102-108) This section describes the development of atomic models and the structure of atoms. Use Target Reading Skills Before you read, preview the diagram of a carbon atom in your textbook. Then, complete the graphic organizer by writing two questions about the diagram. As you read, answer your questions.

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