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interactions invol....

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interactions invol....

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interactions invol....

The separation of
solute particles (ΔH_1

Solvent-solvent
Interactions...

-interactions between
the water m....

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calculation of
quantities in chemical
reactions. The
quantities to consider
in a chemical reaction?
On the left side of a
chemical equation, you
have the? On the right
side of a chemical
equation, you have
the? In every chemical
reaction, these two

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things are conserved.

In a majority of chemical reactions, these three things are NOT conserved.

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Bond Model 12.2

Drawing Lewis

Structures □ General

Procedure □ More Than

One Possible Structure

12.3 Resonance

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25 terms. jakeydvd.

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chemistry. law of

conservation of mass.

formula mass. law of

definite proportions.

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limiting reactant. no detectable change in mass occurs during a chemical reaction. the sum of the atomic masses given in the formula of a substan....

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Liquid and from Liquid
to Gas - An

Introduction to
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The Process of
Condensation The
Process of Evaporation

How Evaporation
Causes Cooling Rate of
Evaporation Dynamic
Equilibrium Between

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Liquid and Vapor
Equilibrium Vapor
Pressure.

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Section 12.1 Gases 1.
motion 2. a. small b.
forces c. random d.
elastic; kinetic 3. KE
 $\frac{1}{2} mv^2$ 4.
Temperature 5. true 6.
true 7. false 8. true 9.
true 10. false 11. true
12. false 13. a 14. a 15.

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d 16. d 17. b 18. b 19.
b 20. barometer 21.
Evangelista Torricelli
22. gravity and
atmospheric pressure
23. The atmospheric
pressure has
increased. Section 12
.2 Forces of Attraction
1. intermolecular
forces 2.
intermolecular forces

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this set (33) Carbon.

Organic chemistry is
the study of all
substances containing.

Inorganic chemistry.

The branch of
chemistry that deals
with products without
carbon is called.

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Alcohols Alcohols are

organic compounds

that have a hydroxyl

(-OH)

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12.2 Stoichiometric

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problems. In the

equation that follows

each problem, write in

the space provided the

mole ratio that can be

used to solve the

problem.

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12 - States of Matter

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motion 2. a. small b.

forces c. random d.

elastic; kinetic 3. $KE =$

$\frac{1}{2} mv^2$ 4.

Temperature 5. true 6.

true 7. false 8. true 9.

true 10. false 11. true

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12. false 13. a 14. a 15.
d 16. d 17. b 18. b 19.
b 20. barometer 21.
Evangelista Torricelli
22. gravity and
atmospheric ...

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small b. forces c.
random d. elastic;
kinetic 3.

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